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Phosphorus Carbene and Olefine Analogues

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PHOSPHORUS CARBENE AND OLEFINE ANALOGUES

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A representative of the yet unknown parent iminophosphane is $\operatorname{Bu}^{\mathsf{t}} P = \operatorname{NBu}^{\mathsf{t}} \ \underline{1}$ which has been recognized as an carbenic analogue for some time $^{\mathsf{l}}$. The unusual properties of $\underline{1}$ prompted our investigations of the synthesis of further phosphorus(III)-(p-p) π -bonds systems with an orbital sequence (HOMO= n(P), LUMO= π^* (P=X)) as well as the reaction behaviour of this class of compounds.

The approach of iminophosphanes of type $\underline{1}$ was achieved by chlorosilane elemination or base promoted dehydrochlorination reaction from the aminophosphanes $\underline{2}^2$.

The stability of $\frac{3}{2}$ stronly depends on the steric and electronic demand of the substituents. N-silylated derivatives (R=Bu^t,R'=SiMe₃, SiMe₂Bu^t) are only accessible in the dimeric form $\frac{4}{2}$, while for N-arylated derivatives (R=Bu^t, Prⁱ, CH(SiMe₃)₂; R'=2,4,6-Bu^t₃C₆H₂) no self dimerisation was observed. Borderline cases are the iminophosphanes (R=Bu^t, R'=adamanty, mesityl) which exist in both forms². A similar reaction afforded the first iminophosphane with the >P-P=N-skeleton (Bu^t₂P-P=N-2,4,6-Bu^t₃C₆H₂; δ ³¹P= 570,99; 430Hz; r_{PN} = 158; \star PPN= 106°)³.

In order to get further insight in the reactivity of carbenic iminophosphanes, the reaction of $\underline{\underline{l}}$ was studied in comparison with the isovalent olefinic methylenephosphane, Bu^tP=CHBu^t $\underline{\underline{5}}^4$. The different reaction behaviour of $\underline{\underline{l}}$ and $\underline{\underline{5}}$ was confirmed by (4+1)- vs.

(4+2)-cycloadditions with 2,3-dimethylbutadiene and 1,1-oxidative addition vs.1,2-addition with halogens and carbon tetrachloride⁵.

The kinetically favoured (2+1)-cycloadduct of the iminophosphane $\underline{1}$ rearranges in the presence of acids (\mathbf{H}^+ , \mathbf{BF}_3) to the thermodynamically stable diazadiphosphetidine $\underline{8}$; in the reaction with triflate acid two intermediates, the phosphonium salt $\underline{6}$ and the diazadiphosphetidinium ion $\underline{7}$ can be isolated $\underline{6}$.

adiphosphetidinium ion
$$\underline{7}$$
 can be isolated⁶,

 $R + H$
 $\underline{4} \xrightarrow{H^+} R + NHR'$
 $R'N \xrightarrow{P} NR' \xrightarrow{-H^+} R'N \xrightarrow{P} NR'$
 $\underline{6} \qquad \underline{7} \qquad R'N \xrightarrow{P} R \qquad \underline{8} \qquad \underline{8} \qquad R'N \xrightarrow{P} R \qquad \underline{8} \qquad$

A novel variety of 1-phosphatriazenes $\operatorname{Bu}^{\mathsf{t}}-\operatorname{P=N-NR}_2$ (R=alky1, pheny1, sily1) were synthesized by elimination reactions from corresponding hydrazophosphanes, $\operatorname{Bu}^{\mathsf{t}}\operatorname{P}(\operatorname{C1})-\operatorname{N}(\operatorname{H},\operatorname{SiMe}_3)-\operatorname{NR}_2^5$. Assignment of the order of orbitals $(\operatorname{n}(\operatorname{P}),\pi(\operatorname{P=N}))$ in these compounds as well as in aminoiminophosphanes was carried out, utilizing the relationship between u.v. data $((\operatorname{n-\pi^*})-(\operatorname{\pi-\pi^*}))$ and ionization potentials (i.p. $\operatorname{n}(\operatorname{P})-i.p.\pi(\operatorname{P=N})$). For alkylated aminoiminophosphanes and 1-phosphatriazenes the π orbital is slighly above σ , while in the case of the silylated derivatives the order of orbitals is reversed. This is in accord with the reaction behaviour towards hexafluoroacetone $((2+2)-\operatorname{vs.}(2+1)-\operatorname{cycloaddition}\operatorname{reaction}^7)$.

The ambident reactive behaviour of the iminophosphane system stimulates the synthesis of methylenephosphanes with a frontier orbital sequence (HOMO= n(P), LUMO= $\pi^*(P=C)$). Suitable derivatives for this should be methylenephosphanes with σ -donor substituents at phosphorus (metal fragments) and π -aceptor substituents (silyl-groups).

An entry to these compounds is the reaction of ${^{Me}}_5{^{C}}_5{^{-P=C}}({^{SiMe}}_3)_2 \stackrel{g}{=} {^{with}}$ transition metal complexes, via shift of the cyclopentadienylligand from phosphorus to the metal $^{8-10}$.

$$\frac{10}{2} \text{ Me}_{5} \text{C}_{5} - \text{P} = \text{CR}_{2}^{2} \qquad \frac{\text{hv}}{-2\text{C}} \qquad \text{Cp*(CO)}_{2} \text{Fe-P=CR}_{2} \qquad \frac{11}{2}$$

$$\frac{\text{Fe(CO)}_{5}}{-\text{CO}} \xrightarrow{\text{MeCN)}_{3} \text{(Mo,W)(CO)}_{3}} \qquad \text{Cp*(CO)}_{3} \text{(Mo,W)-P=CR}_{2} \qquad \frac{12}{2}$$

$$\frac{\text{Ni(COD)}_{2}/\text{R'}_{3}\text{P}}{-2\text{COD}} \qquad \text{Cp*(R'}_{3}\text{P)Ni-P=CR}_{2} \qquad \frac{13}{2}$$

$$(\text{R= SiMe}_{3}; \text{Cp*=η}^{5} - \text{C}_{5}\text{Me}_{5})$$

The high nucleophilicity at the phosphorus is evidenced by the spectroscopic investigations (δ^{31} P= 500 - 740 ppm,(n- π *)= 540 - 640 nm), X-ray crystallographic studies (\sharp M-P-C = 123 - 126°) as in the reaction with electrophiles ^{9,11}. The Ni(CO)₄promoted reaction of metallo-methylene-phosphanes is a new approach to metalla-phosphaallenes⁵.

$$\frac{12}{=2} \quad \frac{\text{"Ni(CO)}_{4}}{\text{-CO}} \quad \text{Cp*(CO)}_{2} \text{W=P=CR}_{2}$$

Realizing that the "ligand shift method" might provide access to the metallo-iminophosphane system, $C_5 \text{Me}_5 - P = \text{NBu}^t$ $\underline{14}$ was treated with $(\text{MeCN})_3 \text{Mo}(\text{CO})_3$. However, this reaction proceeds in a 2:1 molar ratio and affords the spirocyclic compound $\underline{16}$, via a metallo-iminophosphane intermediate $\underline{15}$. The structural investigations of $\underline{16}$ is in accord with a transition metal complex of an $\text{aza-}\lambda^3\lambda^3$ -azadiphosphiridine $\frac{12}{12}$ ($r_{\text{MoP}} = 245$, $r_{\text{PN}(\text{CO})} = 175 \text{ pm}$).

Avoiding electrophilic ligands at the metall fragment the approach to metallo-iminophosphanes was accomplished by the reaction of $\frac{14}{2}$ with (R₃P)₂Ni(COD). Based on n.m.r. investigations the primary formed with η^{1} - and η^{2} -coordinated complexes $\underline{17}$, $\underline{18}$ rearrange with elemination of phosphane to the novel metallo-iminophosphanes $\underline{19}^{10}$.

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